

## “ Reagent Explosion ” : an Efficient Method to Increase Library Size and Diversity.

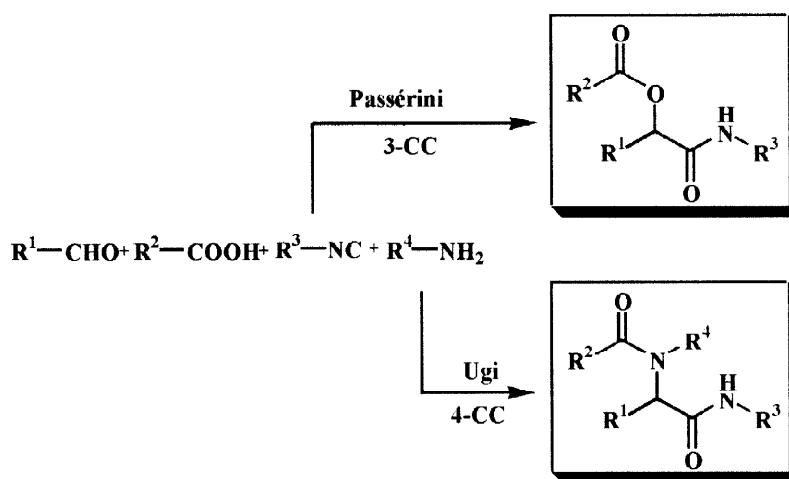
H. Bienaymé

Rhône-Poulenc Technologies, C.R.I.T. BP 62, 69192 St-Fons Cedex, France

Received 12 March 1998; accepted 29 March 1998

**Abstract :** Methyl- $\beta$ -(N,N-dialkylamino)- $\alpha$ -isocyanoacrylates, readily prepared from the reaction between various secondary amines, N-formylimidazole diethylacetal and methyl isocyanoacetate, can be used to increase the diversity of the isonitrile input in multi-component condensation, such as the Passérini reaction. © 1998 Published by Elsevier Science Ltd. All rights reserved.

Multi-component condensations (M-CC), a former relatively confidential reaction class, have recently emerged as one of the most powerful tool for the generation of large collections of molecules (libraries) in combinatorial synthesis.<sup>1</sup> This renewed interest originates from the high convergence, and often relatively simple reaction protocol, of such chemical processes : starting from only 20 chemically distinct building blocks for each reactant, 8000 ( $=20^3$ ) and 160000 ( $=20^4$ ) adducts are theoretically accessible in one step using respectively a three and a four-component condensation. This is illustrated with the Ugi and Passérini M-CC which have recently found widespread use in high throughput synthesis (scheme 1).<sup>2,3</sup>



Scheme 1

However, to reach the kind of figure required for the generation of original « leads » ( $10^3$  to  $10^5$  molecules), each reactant class should be well represented (both in term of number and structural diversity). If this is true for amines, carboxylic acids and aldehydes, isonitriles however remain scarce, as only about 12-15 are commercially available, most of them being aliphatic.

Herein we report our investigation on the Passérini reaction, and our solution to provide structurally diverse isonitriles in a straightforward manner.

\*e-mail : hugues.bienayme@rhone-poulenc.com



step: the isonitrile ester function, the secondary amine and the acylimidazole acetal (the « reagent explosion » principle).

Entry	R <sup>3</sup> R <sup>4</sup> N-	Yield <sup>a</sup> (%)	Entry	R <sup>3</sup> R <sup>4</sup> N-	Yield <sup>a</sup> (%)
1		1a 82	10		1j 63
2		1b 35	11		1k 74
3		1c 30	12		-
4		1d 33	13		-
5		1e 55	14		-
6		1f 33	15		-
7		1g 41	16		-
8		1h 58	17		-
9		1i 58			

a) Isolated yields, products were characterised by all usual means.

**Table 1**

These new reagents were evaluated in the Passérini reaction. They all gave the corresponding adducts in high yields as exemplified in scheme 2. Interestingly, dipolar aprotic solvents such as DMF and DMSO were found superior in this reaction, whereas the commonly employed methanol was not suitable. Importantly, isocyanide stereochemistry was retained throughout this MCC, yielding again pure *Z*-isomeric adducts (racemic).

As both the **Bredereck** and the **Passérini** reaction were conducted in the same solvent, we envisioned the possibility to carry out these two steps in « one-pot », according to the principle of M-CC union.<sup>7</sup> This was accomplished by heating (80°C) for 30 min. a mixture of N-formyl diethylacetal (1.5 eq.), methyl isocyanoacetate (1.0 eq.), morpholine (1.0 eq.) and benzoic acid (1.5 eq.) in DMF, followed by cooling at 25°C and adding cyclohexane carboxaldehyde (2.0 eq.). Adduct **2b** was thus obtained in a fair crystallised yield of 30%. This solution, though very efficient, was not retained for our library production.

Finally, a 4620-members library was prepared in solution (DMF) from 20 different carboxylic acids, 21 aldehydes (mainly aliphatic and electron-deficient aromatic) and the former 11 isonitriles, and screened with some success in various biological assays.

In conclusion, we have shown that the “ reagent explosion ” strategy offers definitive advantage for the rapid construction of combinatorial libraries using M-CC, when one of the reagent input is not readily available or not diverse enough.

**Acknowledgments:** We thank our colleagues at the analytical department (M. Lanson, L. Godde, J.L. Dumoulin and J. Guillaud-Saumur) for their accurate and dedicated contribution, and Rhône-Poulenc Tech. for permission to publish these results.

#### References and notes :

- Curr. Op. Chem. Biol.* **1997**, *1*, 3-145. Hermkens, P.H.H. ; Ottenheijm, H.C.J. ; Rees, D.C. *Tetrahedron*, **1997**, *53*, 5643-5678. Balkenhohl, F. ; von den Bussche-Hünnefeld, C. ; Lansky, A. ; Zechel, C. *Angew. Chem. Int. Ed. Engl.* **1996**, *35*, 2288-2337. Thompson, L.A. ; Ellman, J.A. *Chem. Rev.* **1996**, *96*, 555-600. Gordon, E.M. ; Barrett, R.W. ; Dower, W.J. ; Fodor, S.P.A. ; Gallop, M.A. *J. Med. Chem.* **1994**, *37*, 1233-1251 and 1385-1401.
- Armstrong, R.W. ; Combs, A.P. ; Tempest, P.A. ; Brown, S.D. ; Keating, T.A. *Acc. Chem. Res.* **1996**, *29*, 123-131.
- Ugi M-CC have been extensively used both in solution and on solid phase : Hulme, C. ; Morissette, M.M. ; Volz, F.A. ; Burns, C.J. *Tetrahedron Lett.* **1998**, *39*, 1113. Ugi, I. ; Goebel, M. ; Gruber, B. ; Heilingbrunner, M. ; Heiß, C. ; Hörl, W. ; Kern, O. ; Starnecker, M. *Res. Chem. Intermed.* **1996**, *22*, 625. Harriman, G.C.B. *Tetrahedron Lett.* **1997**, *38*, 5591. Rossen, K. ; Sager, J. ; DiMichele, L.M. *Tetrahedron Lett.* **1997**, *38*, 3183. Short, K.M. ; Mjalli, A.M.M. *Tetrahedron Lett.* **1997**, *38*, 359. Sutherland, D.P. ; Stark, T.M. ; Hughes, R. ; Armstrong, R.W. *J. Org. Chem.* **1996**, *61*, 8350. Ugi, I. ; Demharter, A. ; Hörl, W. ; Schmid, T. *Tetrahedron* **1996**, *52*, 11657. Demharter, A. ; Hörl, W. ; Eberhardt, H. ; Ugi, I. *Angew. Chem. Int. Ed. Engl.* **1996**, *35*, 173. Keating, T.A. ; Armstrong, R.W. *J. Org. Chem.* **1996**, *61*, 8935. Keating, T.A. ; Armstrong, R.W. *J. Am. Chem. Soc.* **1996**, *118*, 2574. Mjalli, A.M.M. ; Sarshar, S. ; Baiga, T.J. *Tetrahedron Lett.* **1996**, *37*, 2943. Strocker, A.M. ; Keating, T.A. ; Tempest, P.A. ; Armstrong, R.W. *Tetrahedron Lett.* **1996**, *37*, 1149. Zhang, C. ; Moran, E.J. ; Woivode, T.F. ; Short, K.M. ; Mjalli, A.M.M. *Tetrahedron Lett.* **1996**, *37*, 751.
- Tempest, P.A. ; Brown, S.D. ; Armstrong, R.W. *Angew. Chem. Int. Ed. Engl.* **1996**, *35*, 640.
- Herdeis, C. ; Dimmerling, A. *Arch. Pharm.* **1986**, *319*, 473. Schöllkopf, U. ; Porsch, P-H. ; Lau, H-H. *Liebigs Ann. Chem.* **1979**, 1444.
- Its preparation is straightforward from imidazole and trimethyl (or triethyl) orthoformate : Brown, R.S. *J. Org. Chem.* **1980**, *45*, 4038. N-Formyltriazole diethylacetal was also prepared and evaluated, but didn't proved superior to its imidazole analog.
- See for instance : Dömling, A. ; Ugi, I. *Angew. Chem. Int. Ed. Engl.* **1993**, *32*, 563.